

Ethyl 4-acetoxy-6-(dimethylamino)-2-naphthoate

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Key indicators

Single-crystal X-ray study

 $T = 298$ KMean $\sigma(\text{C}-\text{C}) = 0.002$ Å R factor = 0.042 wR factor = 0.135

Data-to-parameter ratio = 14.6

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

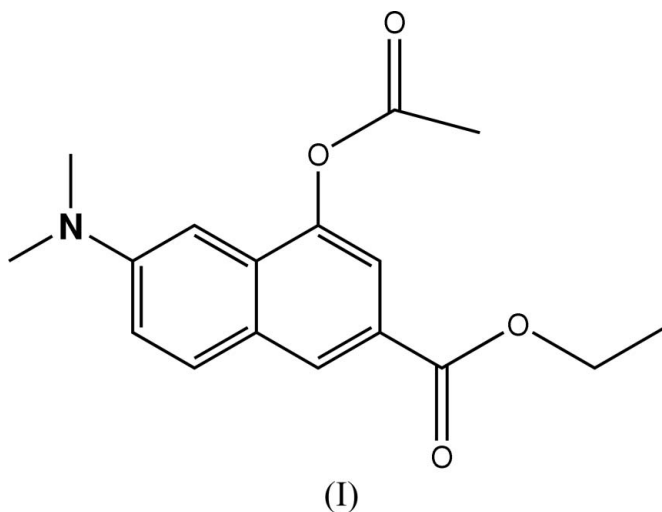
The title compound, $\text{C}_{17}\text{H}_{19}\text{NO}_4$, was synthesized by the Stobbe condensation reaction of *p*-dimethylaminobenzaldehyde and diethyl succinate, followed by cyclization of the Stobbe product. The crystal structure is stabilized by intermolecular $\text{C}-\text{H} \cdots \pi$ interactions.

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Comment

The title compound, (I), is an intermediate in the preparation of photochromic compounds (Clarke *et al.*, 2001). We report here the molecular structure of (I), which is shown in Fig. 1.



The molecule is nearly planar except for the carboxylate groups and some of the hydrogen atoms. $\text{C}-\text{H} \cdots \pi$ interactions occur between $\text{C}17-\text{H}18$ and the $\text{C}1-\text{C}4/\text{C}9/\text{C}10$ ring, such that the distance between $\text{H}18$ and the ring centroid (C_g) is 2.97 Å, while the $\text{C}-\text{H} \cdots C_g^i$ angle is 144° [symmetry code: (i) $x, 1 + y, z$].

Experimental

The title compound was prepared according to the procedures described by Tanaka *et al.* (2000) and Clarke *et al.* (2001), which involved the reaction of *p*-dimethylaminobenzaldehyde (4.47 g, 0.03 mol) and diethyl succinate (7.83 g, 0.045 mol) in the presence of $t\text{BuOK}$ (6.72 g, 0.06 mol) at room temperature. The reaction mixture was neutralized with dilute HCl and extracted with ethyl acetate. Removal of the EtOAc afforded a red-brown solid. A solution of this solid with anhydrous sodium acetate (2.46 g, 0.03 mol) in acetic anhydride (100 ml) was refluxed for 6 h. The crystals formed were isolated by filtration. Yellow single crystals suitable for X-ray data collection were obtained by recrystallization from EtOAc -hexane (1:3 v/v).

Crystal data

$C_{17}H_{19}NO_4$
 $M_r = 301.33$
 Monoclinic, $P2_1/c$
 $a = 12.376 (3) \text{ \AA}$
 $b = 7.0410 (16) \text{ \AA}$
 $c = 17.945 (6) \text{ \AA}$
 $\beta = 97.853 (12)^\circ$
 $V = 1549.1 (7) \text{ \AA}^3$

$Z = 4$
 $D_x = 1.292 \text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation
 $\mu = 0.09 \text{ mm}^{-1}$
 $T = 298 (2) \text{ K}$
 Prism, yellow
 $0.51 \times 0.48 \times 0.32 \text{ mm}$

Data collection

Rigaku R-AXIS RAPID
 diffractometer
 ω scans
 Absorption correction: none
 20499 measured reflections

2916 independent reflections
 2326 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.025$
 $\theta_{max} = 25.6^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.042$
 $wR(F^2) = 0.135$
 $S = 1.10$
 2916 reflections
 200 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0701P)^2 + 0.2653P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.27 \text{ e \AA}^{-3}$
 $\Delta\rho_{min} = -0.22 \text{ e \AA}^{-3}$
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.019 (3)

The methyl H atoms were constrained to an ideal geometry, with C—H distances of 0.96 Å and $U_{iso}(H) = 1.5U_{eq}(C)$, but each group was allowed to rotate freely about its C—C bond. All other H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C—H distances of 0.96 Å for aromatic and 0.97 Å for methylene H atoms, and with $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSD and Rigaku, 2002); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure:

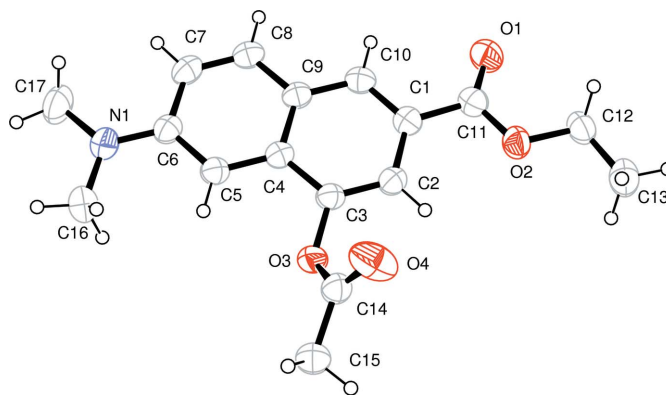


Figure 1
 The molecular structure and atom-labelling scheme for (I), showing 40% probability displacement ellipsoids.

SHELXL97 (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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